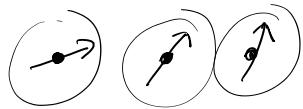
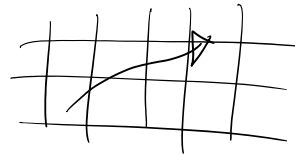


## Lagrangian Method



## Eulerian



Physics : Equation of Motion

$$\underline{F} = m \underline{a}$$

$\delta V$  small "parcel" of the fluid

$$\underline{F} = \rho \delta V \frac{d\underline{u}}{dt}$$

fluid velocity with respect to the moving frame of reference.

$$\frac{d\underline{u}}{dt} = \text{"Time derivative following the motion"} \equiv \frac{\underline{u}(\underline{x} + \underline{u}\delta t, t + \delta t) - \underline{u}(\underline{x}, t)}{\delta t}$$

$$\frac{dF}{dt} = \frac{1}{\delta t} \left\{ \underbrace{F(\underline{x} + \underline{u}\delta t, t + \delta t) - F(\underline{x}, t)}_{\text{Taylor expand}} \right\}$$

$$\left\{ \cancel{F(\underline{x}, t)} + \frac{\partial F}{\partial x} \underline{u}\delta t + \frac{\partial F}{\partial t} \delta t - \cancel{F(\underline{x}, t)} \right\}$$

$$\boxed{\frac{dF}{dt} = (\underline{u} \cdot \nabla) F + \frac{\partial F}{\partial t}}$$

convective derivative

$$\frac{d\underline{u}}{dt} = \underline{u} \cdot \nabla \underline{u} + \frac{\partial \underline{u}}{\partial t}$$

$$\rho \delta V \frac{d\underline{u}}{dt} = \rho \delta V \underline{g}$$

No. We need to include the contribution of pressure

$$\frac{d\underline{u}}{dt} = \underline{g} - \frac{\nabla P}{\rho}$$

(can do the derivation later)

What variables do we have,  $\underline{u}$ ,  $P$  and  $\rho$

5 variables

We have only 3 equations of motion in  $\frac{d\underline{u}}{dt}$  ...

1. Continuity equation : conservation of mass.
2. Energy equation : ✓



Continuity Eq.

1.  $\frac{\partial \rho}{\partial t} + \nabla \cdot (\rho \underline{u}) = 0$  Continuity Eq.

$\underline{u} \cdot \nabla \rho + \rho \nabla \cdot \underline{u}$

$$\frac{d\rho}{dt} + \rho \nabla \cdot \underline{u} = 0$$

Lagrangian Continuity Equation.

2. Conservation of Energy

$$\frac{de}{dt} = - \left( \frac{P}{\rho} \right) \nabla \cdot \underline{u}$$

e: Specific Internal energy  
per unit mass

Total energy is given by:

$$E = \rho \cdot \left( \frac{1}{2} \underline{u} \cdot \underline{u} + e \right)$$

What variables?  $\underline{u}, \rho, e, P$  6 variables and 5 equations



$PV = nRT$  Ideal Gas EOS

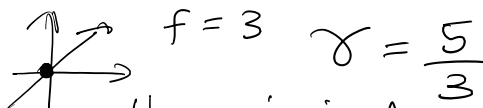
$$e = \frac{P}{\rho(\gamma - 1)}$$

Ideal Gas EOS

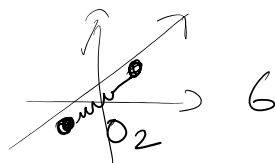
$$\gamma = \frac{f + 2}{f}$$

f is the number of degrees of freedom

Monoatomic



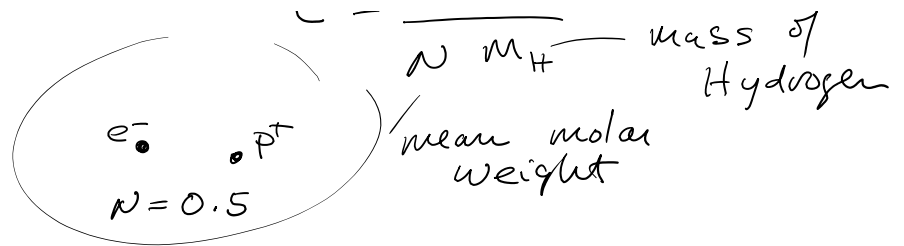
He, ionized hydrogen



Boltzmann constant

$$e = \frac{k_B T}{N m_H}$$

mass of Hydrogen



SPT - use particles to follow the flow and consider various integrals in the calculation of the conservation laws.

$$A_{\underline{r}}(\underline{r}) = \int_{\text{all space}} A(\underline{r}') W(\underline{r}-\underline{r}'; h) d\underline{r}'$$

interpolant  $\nearrow$   
 all space  $\nearrow$  "kernel"  
 approximate via a sum  $\nearrow$

1.  $\int W d\underline{r} = 1$
2.  $\lim_{h \rightarrow 0} W(\underline{r}-\underline{r}'; h) = \delta(\underline{r}-\underline{r}')$   
 limit of a gaussian

$$A_s(\underline{r}) = \sum_b m_b \frac{A_b}{\rho_b} W(\underline{r}-\underline{r}_b; h)$$

$$* \left[ \rho_s(\underline{r}) = \sum_b m_b W(\underline{r}-\underline{r}_b; h) \right]$$

We also need to approximate (or interpolate) gradients of functions:

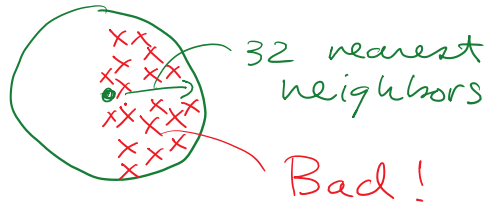
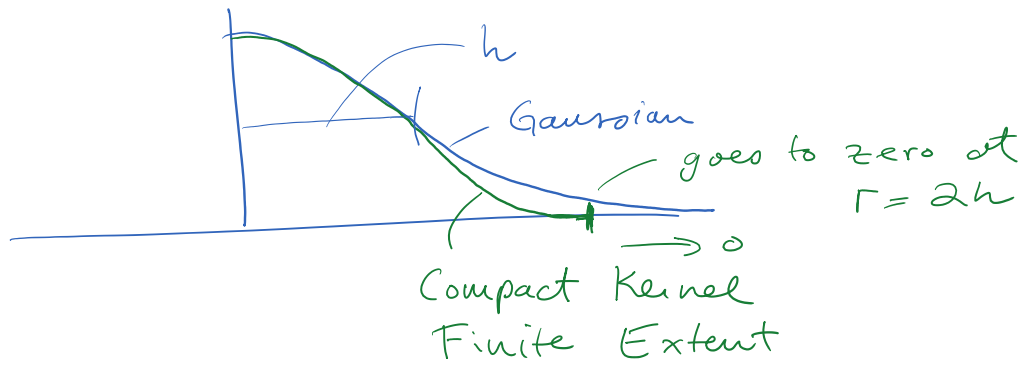
$$\nabla A_s(\underline{r}) = \sum_b m_b \frac{A_b}{\rho_b} \nabla W(\underline{r}-\underline{r}_b; h)$$

can calculate in advance

Instead (higher accuracy) we use:

$$\rho \nabla A = \nabla(\rho A) - A \nabla \rho$$

Kernels: Gaussian  $W(x, h) = \frac{1}{h\sqrt{\pi}} e^{-\left(\frac{x^2}{h^2}\right)}$



Monahan Kernel: "cubic spline kernel"

$$W(r; h) = \frac{\sigma}{h^d} \begin{cases} 6\left(\frac{r}{h}\right)^3 - 6\left(\frac{r}{h}\right)^2 + 1, & 0 \leq \frac{r}{h} < \frac{1}{2} \\ 2\left(1 - \left(\frac{r}{h}\right)\right)^3, & \frac{1}{2} \leq \frac{r}{h} \leq 1 \\ 0, & \frac{r}{h} > 1 \end{cases}$$

$d$  is dimension

Normalizing Constant

$$\sigma = \begin{cases} 4/3 & \text{in 1-D } d=1 \\ 40/7\pi & \text{in 2-D} \\ 8/\pi & \text{in 3-D} \end{cases}$$

$\nabla W$ ?